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Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

| | | | |
|------------------------------|--|---------------------------------------|--|
| Office Action Summary | Application No. 10/763,951 | Applicant(s) JOHNSON ET AL. | |
| | Examiner Matthew J. Merkling | Art Unit 1764 | |

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 25 June 2007.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 1-7 and 9-24 is/are pending in the application.
- 4a) Of the above claim(s) _____ is/are withdrawn from consideration.
- 5) ☐ Claim(s) _____ is/are allowed.
- 6) ☒ Claim(s) 1-7 and 9-24 is/are rejected.
- 7) ☐ Claim(s) _____ is/are objected to.
- 8) ☐ Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on _____ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

- 12) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All b) ☐ Some * c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
 2. ☐ Certified copies of the priority documents have been received in Application No. _____.
 3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

- | | |
|--|---|
| 1) <input checked="" type="checkbox"/> Notice of References Cited (PTO-892) | 4) <input type="checkbox"/> Interview Summary (PTO-413) Paper No(s)/Mail Date. _____ |
| 2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948) | 5) <input type="checkbox"/> Notice of Informal Patent Application |
| 3) <input type="checkbox"/> Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date _____ | 6) <input type="checkbox"/> Other: _____ |

DETAILED ACTION

Claim Rejections - 35 USC § 103

1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

2. Claims 1-3, 9, 10 and 21 are rejected under 35 U.S.C. 103(a) as being unpatentable over Becker et al (US 2002/0006368) in view of Zhou (US 6,500,969) and Li et al. (US 6,782,892).

Regarding claims 1, 9 and 10, Becker discloses:

A fluidized-bed oxidation reactor comprising:

a chamber (Fig. 1 (1)) defining a hollow interior region and having a lower surface (4);

a first input (6) for introducing a gas into the hollow interior region;

a plurality of particles (2) within the hollow interior region and located on the lower surface (4), and;

a fluidizing input (10) for introducing a fluidizing material into the hollow interior region (gas, paragraph 32

Art Unit: 1764

lines 4-5), said fluidizing input having an outlet directed at the lower surface of the chamber (see Fig. 1 (10), paragraph 26 lines 3-5).

Regarding the claimed limitation:

"wherein the introduction of the fluidizing material directed at the lower surface fluidizes at least a portion of the catalyst nanoparticles located on the lower surface to create a gaseous dispersion of catalyst nanoparticles that reacts with the contaminated gas to produce a decontaminated gas."

This does not add any structure to the claimed apparatus and simply states the intended use of said apparatus. MPEP §2115.

Becker fails to teach the plurality of catalysts as being nanoparticles comprising metal.

Zhou also discloses an oxidation process (as does Becker) and the type of catalyst used in said oxidation process.

Zhou teaches nanoparticles comprising platinum (col. 8 lines 33-43) being utilized as the catalyst in an oxidation reaction in order to ensure high activity and selectivity of desired oxidation products (col. 5 lines 34-43).

It would have been obvious to one of ordinary skill in the art at the time of the invention to use the

Art Unit: 1764

nanoparticles comprising platinum of Zhou in the fluidized bed oxidation reactor of Becker in order to ensure high activity and selectivity of the desired oxidation products.

Modified Becker teaches nano-sized oxidation catalyst in the range of 0.5 to 100nm (see Zhou, claim 15), but is silent on particle sizes in the range of 15-25nm.

Li also discloses a nanosized oxidation catalyst.

Li teaches a nanocatalyst used for oxidation with a particle size of 25nm as a preferable way of oxidizing a reactant (col. 8 lines 55-61).

It would have been obvious to one of ordinary skill in the art at the time of the invention to use a particle diameter of 25nm, as in Li, in the catalyst of modified Becker as a known size of nanoparticle to oxidize a reactant.

Regarding claim 2, Becker, further discloses that the nanoparticles will be fluidized by the inlet of gas from the first inlet (paragraph 32 lines 4-8).

Regarding claim 3, Becker further discloses a fluidized-bed chamber comprising a port (Fig. 1, (8)) for the exit of the decontaminated gas out of the hollow interior region (paragraph 35 line 9).

Regarding claim 21, Becker discloses a method of removing contaminants from a contaminated gas comprising:

A fluidized-bed oxidation reactor comprising:

a chamber (Fig. 1 (1)) defining a hollow interior region and having a lower surface (4);

a first input (6) for introducing a gas into the hollow interior region;

a plurality of particles (2) within the hollow interior region and located on the lower surface (4), and;

a fluidizing input (10) for introducing a fluidizing material into the hollow interior region (gas, paragraph 32 lines 4-5), said fluidizing input having an outlet directed at the lower surface of the chamber (see Fig. 1 (10), paragraph 26 lines 3-5).

Becker fails to teach the plurality of catalysts as being nanoparticles.

Zhou also discloses an oxidation process (as does Becker) and the type of catalyst used in said oxidation process.

Zhou teaches nanoparticles being utilized as the catalyst in an oxidation reaction in order to ensure high activity and selectivity of desired oxidation products (col. 5 lines 34-43).

It would have been obvious to one of ordinary skill in the art at the time of the invention to use the nanoparticles of Zhou in the fluidized bed oxidation reactor of Becker in order to ensure high activity and selectivity of the desired oxidation products. (as noted above) and further discloses introducing a recycle/contaminated gas into the hollow interior region (paragraph 32). Becker also discloses introducing a fluidizing material (gas, paragraph 35 lines 6-7) directed at the lower surface (see Fig. 1 (10)).

Modified Becker teaches nano-sized oxidation catalyst in the range of 0.5 to 100nm (see Zhou, claim 15), but is silent on particle sizes in the range of 15-25nm.

Li also discloses a nanosized oxidation catalyst.

Li teaches a nanocatalyst used for oxidation with a particle size of 25nm as a preferable way of oxidizing a reactant (col. 8 lines 55-61).

It would have been obvious to one of ordinary skill in the art at the time of the invention to use a particle diameter of 25nm, as in Li, in the catalyst of modified Becker as a known size of nanoparticle to oxidize a reactant.

Art Unit: 1764

3. Claims 4, 5, 7, 22, 23, and 24 are rejected under 35 U.S.C. 103(a) as being unpatentable over Becker and Zhou as applied to claim 3 above, and further in view of Alford et al. (US 6,887,291) and Ballantine et al. (US 2006/0078771).

Regarding claim 4, the modified Becker discloses all of the claim's limitations, including a second input (Fig. 1 (7)), but does not disclose an input for introducing a backpressure pulse of gaseous material.

Alford also discloses a filter device for removing nanomaterials from gas streams using a gas permeable separating device (Fig.1 (2), see Abstract).

Alford teaches a second input (5) for introducing a backpressure pulse (pulse jet) of gaseous material into a hollow interior region (10) (col. 7 lines 59-67) in order to clean a filter (col. 7 lines 43-55).

It would have been obvious to one of ordinary skill in the art at the time of the invention to use the backpressure pulse input of Alford with the fluidized bed oxidation reactor with nanoparticles of the modified Becker in order to clean the filter that is used to separate the nanoparticles from the gas stream with the second input.

Regarding claim 5 and 22 and 23, the modified Becker discloses all of the claim's limitations but fails to teach:

a gas permeable separation device in communication with said port and the exit of gas from the hollow interior region through the gas permeable separation device,

the gas permeable separation device also in communication with the second input, and

an entrance of the backpressure pulse into the hollow interior region displacing collected catalyst nanoparticles.

Alford also discloses a filter device for removing nanomaterials from gas streams using a gas permeable separating device (Fig.1 (2), see Abstract).

Alford teaches a gas permeable separation device (filter, 2) in communication with a hollow interior region (10) and the second input (5) and the entrance for introducing a backpressure pulse (pulse jet) into the hollow interior region (10) displacing collected catalyst nanoparticles (col. 7 lines 43-55). Alford teaches this in order to allow catalyst nanoparticles to be collected by said gas permeable separation device (filter) and to clean said gas permeable separation device of said catalyst nanoparticles (col. 7 lines 35-67).

It would have been obvious to one of ordinary skill in the art at the time of the invention to use the gas permeable separation device (in communication with the second input) and the entrance of the backpressure pulse into the hollow interior region to displace the collected nanoparticles of Alford, with the fluidized bed oxidation reactor of the modified Becker in order to allow catalyst nanoparticles to be collected by said gas permeable separation device and to clean said gas permeable separation device of said catalyst nanoparticles.

Regarding claims 7 and 24, modified Becker, as discussed in claims 4 and 23 above, discloses a vessel with inlet (process inlet, backpressure pulse) and outlet

Art Unit: 1764

lines (process outlet) with controlled valves alternating in the open and closed position (see Alford, as described above). However, modified Becker does not teach synchronizing the backpressure pulse valve with the first input valve.

Ballantine also discloses a series of controlled valves introducing and extracting fluids from a process vessel.

Ballantine teaches a valve synchronization process of closing an inlet valve (402) connected to a vessel (412) when a second inlet valve (401) is opened in order to prevent backflow through the first process valve (paragraph 42)

It would have been obvious to one of ordinary skill in the art at the time of the invention to change the control scheme of the valve controller in modified Becker to synchronize the valve opening of the second inlet with the valve opening of the first inlet such that when the second inlet valve opens, the first inlet valve closes in order to prevent back flow through the first inlet valve, as is taught by Ballantine.

4. Claim 6 is rejected under 35 U.S.C. 103(a) as being unpatentable over Becker and Zhou as applied to claim 1 above, and further in view of Goswami (US 5,933,702).

Regarding claim 6, the modified Becker discloses all of the claims limitations, as discussed with respect to claim 1 above, but does not teach a humidifier in communication with the first input (gas inlet).

Art Unit: 1764

Goswami also discloses a photocatalytic/oxidation reactor for reacting a gas to remove contaminants via oxidation.

Goswami discloses a humidifier (Fig. 1 (50)) on the gas inlet (18) to a photocatalytic/oxidation reactor (21) in order to provide the correct relative humidity for the complete oxidation and destruction of a microorganism in the photocatalytic/oxidation reactor (col. 7 line 60 - col. 8 line 4).

It would have been obvious to one of ordinary skill in the art at the time the invention was made to use the humidifier and photocatalytic/oxidation reactor of Goswami with the fluidized bed oxidation reactor of Becker in order to ensure the correct humidity for the complete oxidation and destruction of said microorganisms.

5. Claims 11 and 12 are rejected under 35 U.S.C. 103(a) as being unpatentable over Becker and Zhou as applied to claim 1 above, and further in view of Wu (US 2002/0187082).

Regarding claims 11 and 12, the modified Becker discloses all of the claim's limitations as discussed in claim 1 above, but fails to teach an ultraviolet light as well as the ultraviolet light within the hollow interior region of the chamber.

Art Unit: 1764

Wu teaches a photocatalytic/oxidation reactor (Fig. 3(a) (315)) which uses photocatalysts to treat polluted air.

Wu also teaches an ultraviolet light (320) in order to facilitate chemical reactions in photocatalysis (paragraph 8, lines 1-4). Wu further teaches said ultraviolet light being positioned within the hollow interior of the chamber (315). It is well known in the art that positioning the ultraviolet light inside the reactor or chamber maximize the exposure of the photocatalyst or the photoactive material, as is shown by Sanderson (US 2005/0079124, paragraph 113).

It would have been obvious to one of ordinary skill in the art at the time of the invention to use the internally positioned ultraviolet light of Wu with fluidized bed oxidation reactor of the modified Becker in order to facilitate chemical reactions in photocatalysis and maximize the exposure of the photocatalyst.

6. Claims 11 and 13 is rejected under 35 U.S.C. 103(a) as being unpatentable over Becker and Zhou as applied to claim 1 above, and further in view of Sato (US 6,812,470).

Regarding claims 11 and 13, the modified Becker discloses all of the claims limitations, as discussed in claim 1

Art Unit: 1764

above, but fails to teach the ultraviolet light positioned outside of the chamber/reactor.

Sato also discloses a photocatalytic/oxidation reactor chamber (Fig. 2 (50)).

Sato teaches an ultraviolet light (80) positioned outside of the reactor chamber in order to facilitate preventing the ultraviolet light from overheating by using a fan blowing external air (col. 5 lines 18-24).

It would have been obvious to one of ordinary skill in the art at the time of the invention to use the external ultraviolet light of Sato with the photocatalytic/oxidation reactor of the modified Becker in order to facilitate prevention of the ultraviolet light overheating by using a fan blowing external air.

7. Claim 14 is rejected under 35 U.S.C. 103(a) as being unpatentable over Becker and Zhou and Wu as applied to claim 11 above, and further in view of Goswami (US 5,933,702).

Regarding claim 14, the modified Becker discloses all of the claims limitations, as discussed in claim 11 above, but does not teach a humidifier in communication with the first input (gas inlet).

Art Unit: 1764

Goswami teaches a photocatalytic/oxidation reactor for reacting a gas to remove contaminants.

Goswami also teaches a humidifier (Fig. 1 (50)) on the gas inlet (18) to a photocatalytic/oxidation reactor (21) in order to provide the correct relative humidity for the complete oxidation and destruction of a microorganism in the photocatalytic reactor (col. 7 line 60 - col. 8 line 4).

It would have been obvious to one of ordinary skill in the art at the time the invention was made to use the humidifier and photocatalytic/oxidation reactor of Goswami with the fluidized bed photocatalytic/oxidation reactor of the modified Becker in order to ensure the correct humidity for the complete oxidation and destruction of said microorganisms.

8. Claim 15 is rejected under 35 U.S.C. 103(a) as being unpatentable over Becker and Zhou and Wu as applied to claim 11 above, and further in view of Sherman (US 6,653,356).

Regarding claim 15, the modified Becker teaches all of the claim's limitations as discussed in claim 11 above, but does not disclose groups included in the photocatalytic material.

Sherman teaches the production of photocatalytic nanoparticles and describes uses therein, such as its anti-microbial (catalytic oxidation) properties.

Sherman also teaches that a type of photocatalytic material to be used on nanoparticles is titanium dioxide in order to save costs and leverage anti-microbial effects in the presence of ultraviolet light (Abstract and paragraph 4).

It would have been obvious to one of ordinary skill in the art at the time of the invention to use the titanium dioxide nanoparticles of Sherman with the oxidation/photocatalytic reactor and the nanoparticles of the modified Becker in order to save costs and leverage antimicrobial effects in the presence of ultraviolet light.

9. Claims 16 and 17 are rejected under 35 U.S.C. 103(a) as being unpatentable over Becker and Zhou and Wu as applied to claim 11 above, and further in view of Wei et al. (US 2005/0129591).

Regarding claims 16 and 17, the modified Becker discloses all of the claims limitations as discussed in claim 11 above, but does not teach a nanoparticle comprising a metal oxide and a co-catalyst.

Art Unit: 1764

Wei discloses a photocatalyst for air quality treatment (see title).

Wei teaches a nanoparticle photocatalyst that contains a metal oxide (titanium oxide) in order to destroy contaminants in an air purifier (paragraph 3 lines 1-2). Wei also teaches a co-catalyst (gold) in order to act together with the titanium dioxide as an effective thermocatalyst for room temperature oxidation of carbon monoxide to carbon dioxide (paragraph 4).

It would have been obvious to one of ordinary skill in the art at the time the invention was made to use the metal oxide photocatalyst and gold co-catalyst of Wei with the fluidized photocatalytic/oxidation reactor of Becker in order to destroy air contaminants and oxidize carbon monoxide to carbon dioxide at room temperature.

10. Claims 18-20 are rejected under 35 U.S.C. 103(a) as being unpatentable over Becker and Zhou as applied to claim 1 above, and further in view of Sigai (US 4,585,673).

Regarding claims 18-20, the modified Becker discloses all of the claims limitations as discussed in claim 1 above, but does not teach a means for agitating the catalyst nanoparticles in the hollow interior region.

Art Unit: 1764

Sigai also discloses a fluidized bed chamber (Fig. 1 (15)).

Sigai teaches an agitation/vibrating/shaking system (Fig. 1 (17,19)) in order to fluidize a suspended solid (in this case, phosphor powder) and improve the expansion of the fluidized bed (col. 4 lines 46-50).

It would have been obvious to one of ordinary skill in the art at the time of the invention to use the agitation/shaking/vibrating means of Sigai with the fluidized bed oxidation reactor of Becker in order to fluidize the suspended solid and improve the expansion of the fluidized bed.

Response to Arguments

35 USC §112 Rejections

11. The rejection of claim 10 under 35 USC §112 has been withdrawn in light of the amendments.
12. The rejections of claims 7, 8, 10, 16, 17 and 24 under 35 USC §112 have been withdrawn in light of the amendments.

35 USC §103(a) Prior Art Rejections

13. Applicant's arguments filed 6/25/07 have been fully considered but they are not persuasive.

Art Unit: 1764

14. Regarding applicants arguments of the prior art rejections of claims 1 and 21:

First, Applicant's argument that Zhou fails to teach an average particle diameter of about 15-25nm have been considered but are moot in view of the new ground(s) of rejection.

In response to applicant's argument that the references fail to show certain features of applicant's invention, it is noted that the features upon which applicant relies (i.e., average particle diameter of 15-25nm) are not recited in the rejected claim(s). Although the claims are interpreted in light of the specification, limitations from the specification are not read into the claims. See *In re Van Geuns*, 988 F.2d 1181, 26 USPQ2d 1057 (Fed. Cir. 1993).

Second, Applicant argues that Zhou does not teach the noble metal catalyst nanoparticles for the same purpose as Applicants' invention (oxidation reaction). The examiner respectfully disagrees as Zhou does teach the noble metal nanoparticles to oxidize hydrogen to produce an intermediate H₂O₂ (see abstract). Examiner identified Zhou's preference of nanoparticles to achieve this feature (col. 5 lines 34-43). Furthermore, in response to applicant's argument that Zhou does not teach the catalyst nanoparticles used for the same purpose, a recitation

Art Unit: 1764

of the intended use of the claimed invention must result in a structural difference between the claimed invention and the prior art in order to patentably distinguish the claimed invention from the prior art. If the prior art structure is capable of performing the intended use, then it meets the claim.

Third and fourth, Applicant argues that Zhou teaches away from using the oxidizing agents disclosed in Becker. The examiner respectfully disagrees. As Applicant points out, Zhou states that such oxidizing agents (O₂) are dangerous and explosive. However, Becker recognizes these problems with O₂ as well and rectifies the situation by using detectors and shutting down the oxygen flow if the situation becomes dangerous (paragraph 20). As such, Zhou does not teach away from Becker.

15. Regarding applicants arguments of the prior art rejections of claims 4, 5, 7, 8, 22, 23, and 24:

Applicant argues that Alford does not teach or suggest introducing the back pressure gas pulse back into the hollow interior region. Examiner respectfully disagrees. As Alford illustrates in Fig. 1, a back pressure pulse from jet (20) would undoubtedly pass into the hollow interior region (10) as there is no valve or blockage separating the filter (2) and the hollow interior region (10).

Art Unit: 1764

Applicant also argues that Alford does not teach synchronization between the introduction of gas pulse with the introduction of one or more other gasses into the hollow interior region. In response to applicant's argument that the references fail to show certain features of applicant's invention, it is noted that the features upon which applicant relies (as stated above) are not recited in the rejected claim(s). Although the claims are interpreted in light of the specification, limitations from the specification are not read into the claims. See *In re Van Geuns*, 988 F.2d 1181, 26 USPQ2d 1057 (Fed. Cir. 1993).

Art Unit: 1764

Conclusion

16. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

17. Any inquiry concerning this communication or earlier communications from the examiner should be directed to Matthew J. Merkling whose telephone number is (571)272-9813. The examiner can normally be reached on M-F 8:30-4:30.

Art Unit: 1764

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Glenn Caldarola can be reached on (571)272-1444. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.



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